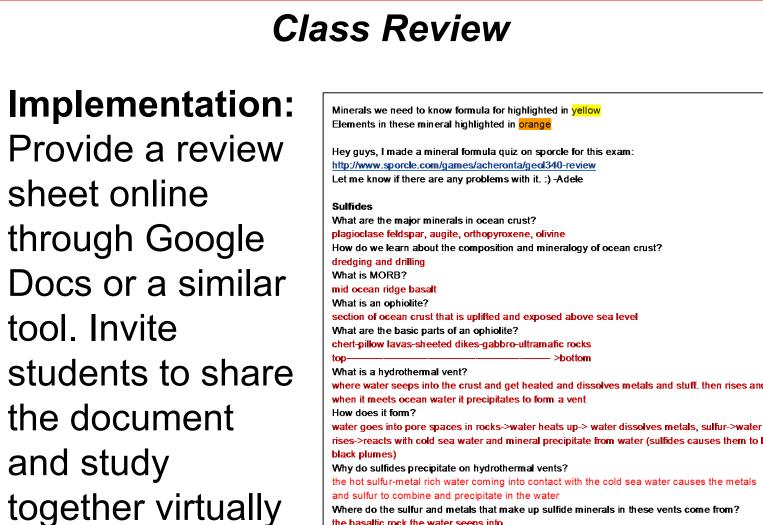
# **Collaborative Documents**

## What it is:

Activities that encourage students to work together in or out of the classroom by using on-line collaborative tools. Many possibilities – 3 examples are shown here. May be combined with an on-line peer review tool (*e.g.* iPeer or PRAZE).



# Wiki Page or Web Page

### Implementation:

- Give an assignment with specific instructions and grading rubric. Students work in small groups over course of week.
- Students research and create a wiki or web page that describes topic, includes images, and cites sources.
- Evaluate their creations!

# The assignment should:

 Allow students to investigate and learn information that is outside the prescribed coursework that interests them

Pyrope Mineral formula: Mg<sub>3</sub>Al<sub>2</sub>Si<sub>3</sub>O<sub>12</sub> Crystal system: Isometric Crystal Habit: Dodecahedral; often found as granular

	the basaltic rock the water seeps into		
	What is the nature of chemical bonds in sulfides?		
	chemical bonding in the sulfides is less ionic than oxides, sulfur is significantly less		
	electronegative than oxygen. Bonding is covalent or metallic depending on energy orbitals.		
	Know the mineral formulae for the seven sulfide minerals you have learned in this class:		
	Pyrite(FeS <sub>2</sub> ), Galena(PbS), Sphalerite((Zn,Fe)S), Pyrrhotite(Fe <sub>(1-x)</sub> S), Chalcopyr	rite(CuFeS <sub>2</sub> ),	
	Molybdenite(MoS <sub>2</sub> ), Cinnabar(HgS)		
	What are distinctive characteristics of each? What are common characteristics? Where are each		
	What are distinctive characteristics of each? What are common characteristics?	Where are each?	
	What are distinctive characteristics of each? What are common characteristics? of these minerals found?	Where are each?	
		Where are each?	
	of these minerals found?   SEE CHART   mineral formula   distinctive characteristic common   w	Where are each where they are ound	

#### Presentations

**Implementation:** Ask students to create a presentation using one of the many on-line presentations apps available. Provide a rubric showing how they will be evaluated. Students work together in small groups to prepare a presentation, and present it to the class.

Designed by Rachel Beane, Bowdoin College On the Cutting Edge project Hardness on Mohs scale: 7 - 7.5 Specific gravity: 3.56 gm/cc Color: deep red Luster: vitreous Streak: white Fracture: conchoidal (Sources: <u>1</u>, <u>2</u>)



Small pyrope crystal, showing characteristic 'blood red' color and dodecahedral habit. Scale: crystal is 2.5mm (Mindat).

Pyrope in the metamorphic rock Eclogite. The accompanying green mineral is pyroxene. Image size: 13cm (<u>Sandatlas</u>) A coesite and quartz inclusion in pyrope garnet. Characteristic radial fractures in the pyrope are visible. It was the unlikely presence of nearly pure pyrope that encouraged Chopin to check for coesite. Picture width:

1.3mm (Elements).

#### Interesting Fact:

In 1984, geologist Christian Chopin published a paper describing his sampling of metamorphic coesite and pure pyrope in the Western Alps. His seemingly minor find went on to change the way scientists understood plate tectonics. Previously, diagrams showing the tectonic movement of continental plates had restricted such movement to the normal thickness of the continental crust. Pyrope has a high-pressure stability field, and pyrope-rich garnet had previously only been found in rocks with origins in the mantle. Chopin's findings showed that crustal rocks had penetrated much deeper into the earth than previously believed, traveling down at least 100km. The discovery of this rare pyrope and its inclusions ultimately brought around a new theory of continental tectonics (Elements, European Journal of Mineralogy).

Chopin's original article: <u>Coesite and pure pyrope in high-grade blueschists of the Western Alps:</u> <u>a first record and some consequences</u>.

http://serc.carleton.edu/NAGTWorkshops/undergraduate\_research/collaborate.html